

MCP-003-001537

Seat No.

Third Year B. Sc. (Sem. V) (CBCS) Examination

May / June - 2018

IC - 501 : Dye - 1 & Petrochemicals

Faculty Code: 003

Subject Code: 001537

Time : $2\frac{1}{2}$ Hours]

[Total Marks: 70

Instructions: (1) All the questions are compulsory.

- (2) Figures to the right indicate maximum marks.
- (3) Draw labeled diagram wherever necessary.
- (4) Assume suitable data.
- (5) Question-1 carries 20 marks.
- (6) Question-2 & 3 carries 25 marks each.
- 1 Answer the following questions:

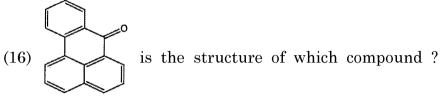
20

- (1) Which catalysts are used in synthesis of ethanol?
- (2) Which of the compound can favor the production of methanol when added in methane stream?
- (3) Reaction is exothermic in nature in manufacturing of styrene. True/False
- (4) In manufacturing of Cumene, Benzene and olefin stream is charged in packed bed reactor in the molar ratio of .
- (5) Bisphenol-A is a monomer for the production of Epoxy resins. True/False
- (6) Adipic acid is a monomer for the production of Nylon-6. True/False
- (7) In Naphtha steam reforming, what is the meaning of LTS?
- (8) To separate i-butane and i-butene, Extractive distillation is used. True/False
- (9) Why N_2 is introduced in manufacturing of acetic acid?
- (10) Carbon monoxide and hydrogen mixture known as _____.

- (11) Give one example of Basic dye.
- (12) 'Chromophore Auxochrome' theory was given by

(13) According to Molecular Orbital theory _____ transition requires highest energy.

- (14) Logwood is an example of Synthetic dye. True/False
- (15) Color fastness can be denoted on the scale from _____ to _____.



- (17) Enlist various Effluent Treatment Plant equipment's.
- (18) Give the structure of Algol yellow.
- (19) Give one function of dispersing agent.
- (20) Enlist various types of dying process.

2 (A) Answer Any Three:

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- (1) Write chemical reaction for ethylene oxide.
- (2) Write properties and chemical reaction for Acetylene.
- (3) Write chemical reaction for manufacturing of Toluene diisocynate.
- (4) Define: (i) Dye (ii) Pigment
- (5) Explain: Difference between Reactive dyes and Direct dyes (any two)
- (6) Give any two function dispersing agent.

(B) Answer Any Three:

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- (1) Discuss Ethylene glycol in brief.
- (2) Write chemical reaction involved in manufacturing of isopropanol (IPA).
- (3) Draw only process flow diagram for BTX Extraction from BTX crude.
- (4) Give reason: p-Amino azo benzene is yellow but in acidic solution it becomes violet.
- (5) Give reason: Ethylene is colorless but β -carotene is orange red.
- (6) Give the synthesis of Disperse Red 4

(C)	Answer Any Two :		10
	(1)	Discuss the manufacturing of Vinyl acetate monomer in detail.	10
	(2)	Explain manufacturing of butadiene in detail.	
	(3)	Give an account of Natural gas steam reforming in detail.	
	(4)	Explain: Molecular Orbital Theory	
	(5)	Give any two synthesis of Indigo.	
(A)	Answer Any Three :		6
	(1)	Write chemical reaction and uses of Carbon disulphide.	
	(2)	Draw scheme for production of CO.	
	(3)	Enlist properties of Acrylonitrile.	
	(4)	Explain: Requisites of a true dye (any four)	
	(5)	Give synthesis of Flavanthrone	
	(6)	Enlist imitations of poor plant layout (any four).	
(B)	Answer Any Three :		9
	(1)	Write properties and chemical reaction of iso-butanol.	
	(2)	Draw only process flow diagram for manufacturing of Acrylic acid.	
	(3)	Write chemical reaction and enlist uses of Caprolactum.	
	(4)	Give reason: Benzene is colorless; p-Nitro benzene is pale yellow while p-Nitro aniline is dark yellow.	
	(5)	Give synthesis of Indanthrene Yellow 4GK.	
	(6)	Explain: Important data for plant scale-up and pilot plant.	
(C)	Answer Any Two :		10
	(1)	Explain manufacturing of Adipic acid in detail.	
	(2)	Explain manufacturing of Ethylene in detail.	
	(3)	Explain: Manufacturing of Indanthrene Rubene-R in detail.	

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(5) Explain: Witt's theory in detail.

(4) Give manufacturing of Reactive Red in detail.